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the quantitative relationships between the amounts of reactants used and the products formed by a chemical reaction. b. Mole -The SI unit used to measure the amount of a substance that contains 6.02×10^{23} atoms of that substance. Stoichiometry Study Guide KEY Chemistry RHS Mr. Moss STUDY GUIDE FOR CONTENT MASTERY CHAPTER 13 Vacuum Pressure exerted by mercury column Atmospheric pressure 760 mm Name Date Class

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the quantitative
relationships between the
amounts of reactants
used and the products
formed by a chemical

reaction. b. Mole -The SI
unit used to measure the
amount of a substance
that contains 6.02×10^{23}
atoms of that substance.

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