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*- Cell Potential \u0026amp; Notation, Redox Half Reactions, Nernst Equation*

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 electrochemical reactions; which deal  
 with the relationship between electrical  
 energy and chemical reactions.  
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 transfer of electrons and are essentially  
 REDOX reactions. Displacement and  
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**Potential \u0026 Notation, Redox**

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reactions in which charged particles

(ions or electrons) cross the interface  
between two phases of matter, typically  
a metallic phase (the electrode) and a  
conductive solution, or electrolyte. A  
process of this kind is known generally  
as an electrode process.

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Electrical Conduction: The substances  
are divided into 4 types depending upon  
their capability of flow of electrons. i)  
Conductors: The Substances which  
allows electricity to pass through them  
are called conductors. Ex :- Metals,  
metal sulphides, acids, alkalis, salt sol.  
and fused salts

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Electrochemistry is that branch of  
chemistry which deals with the study of  
production of electricity from energy  
released during spontaneous chemical  
reactions and the use of electrical  
energy to bring about non-spontaneous  
chemical transformations. Importance of  
Electrochemistry 1. Production of metals  
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