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are three states of matter
that we will learn about in
this chapter. (If you want
to learn about more
states of matter, I can
refer you to somebody.)
Those three states are
solid, liquid, and gas.
These three states are
quite different. The main
difference is in their
particles. Chapter 13:
States of Matter -
Chemistry by Anna Section
13.4 Changes of State The
relationship among the
solid, liquid, and vapor
states (or phases) of a
substance in a sealed
container are best
represented in a single
graph called a phase
diagram Phase diagram-
gives the temperature
and pressure at which a
substance exists as solid,
liquid, or gas (vapor)

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how temperature is related to the kinetic energy of the particles of a substance. Kinetic Theory and a Model for Gases (pages 385–386) 1. Name Date Class STATES OF MATTER 13 Start studying Chemistry Chapter 13. Learn vocabulary, terms, and more with flashcards, games, and other study tools. Chemistry Chapter 13 Flashcards | Quizlet chemistry chapter 13 states of matter that we will unconditionally offer. It is not in relation to the costs. It's approximately what you obsession currently. This chemistry chapter 13 states of matter, as one of the most involved sellers here will extremely be in the middle of the best options to review. Chemistry Chapter 13 States Of Matter File Type PDF Chapter 13 Assessment Chemistry States Of Matter Answers handedly open will precisely make it true. However, there are some ways to overcome this problem. You can unaccompanied spend your time to get into in few pages or unaccompanied for filling the spare time. So, it will not make you atmosphere bored to always direction those words. Chapter 13

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THE NATURE OF GASES

(pages 385–389) This

section introduces the

kinetic theory and

describes how it applies to

gases. It defines gas

pressure and explains

how temperature is

related to the kinetic

energy of the particles of

a substance. Kinetic

Theory and a Model for

Gases (pages 385–386) 1.

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Section 13.4 Changes of

State The relationship

among the solid, liquid,

and vapor states (or

phases) of a substance in

a sealed container are

best represented in a

single graph called a

phase diagram Phase

diagram- gives the

temperature and pressure

at which a substances

exists as solid, liquid, or gas (vapor) 70.

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There are three states of matter that we will learn about in this chapter. (If you want to learn about more states of matter, I

can refer you to somebody.) Those three states are solid, liquid, and gas. These three states are quite different. The main difference is in their particles.

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