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 substance is called a mole • A mole (mol) of a substance is

equivalent to 6.02×10^{23} particles of that substance • The mole was founded by a scientist named Avogadro, and he decided to use the

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 10. Look at Figure 10.16 and Table 10.3. Name three compounds that have an empirical formula of CH. 11. Fill in the labels on the diagram below.
 MICROSCOPIC INTERPRETATION
 MACROSCOPIC INTERPRETATION
 SO₃ molecule 1 mol SO₃ SO₃ composed of composed of S atom and sulfur atoms (1023) oxygen atoms 3 atoms and
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